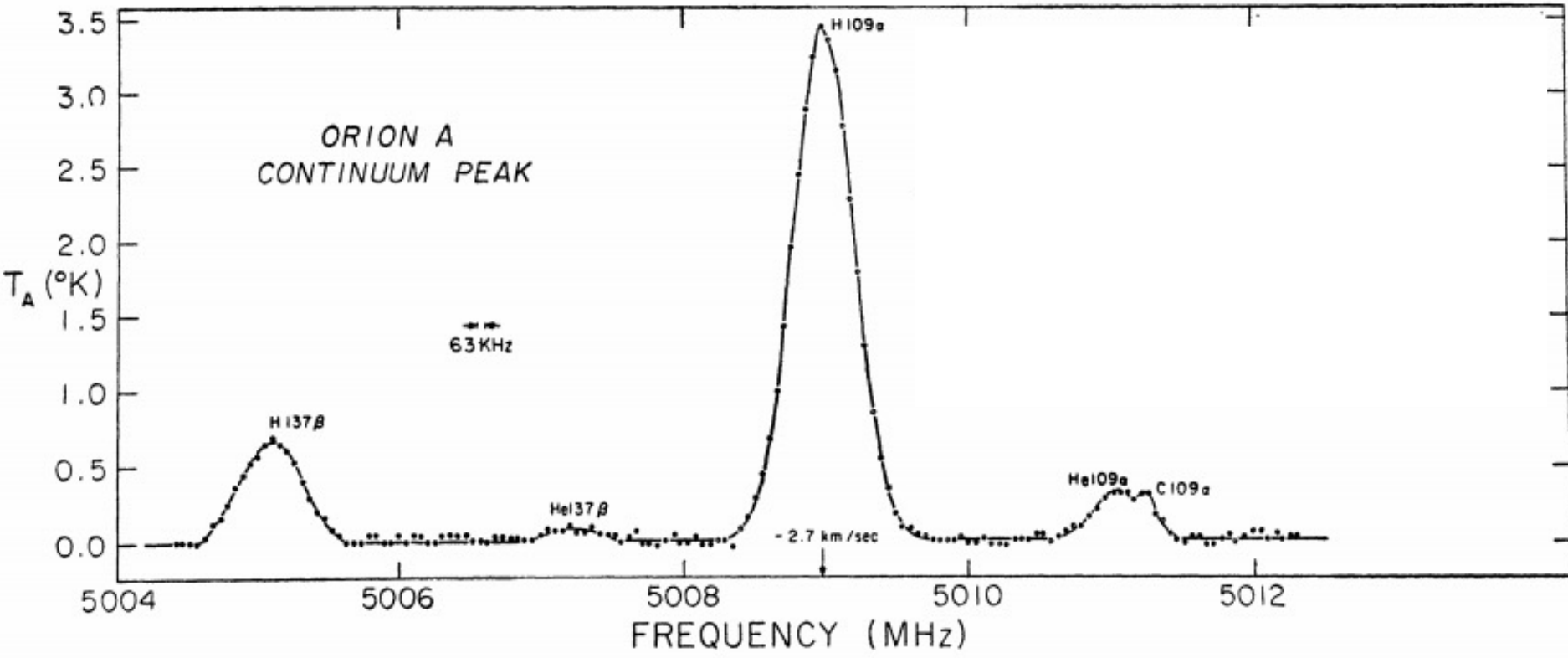


# Spectral Lines

- Uses of spectral lines
- Opacity for spectral lines
- Profiles of spectral lines
- Examples of radio spectral lines

# Spectral Lines in Radio Astronomy

- Up to now we have mostly discussed processes that give rise to radio continuum emission such as the free-free and synchrotron mechanisms
- There are also examples of spectral line emission (and absorption) in radio astronomy

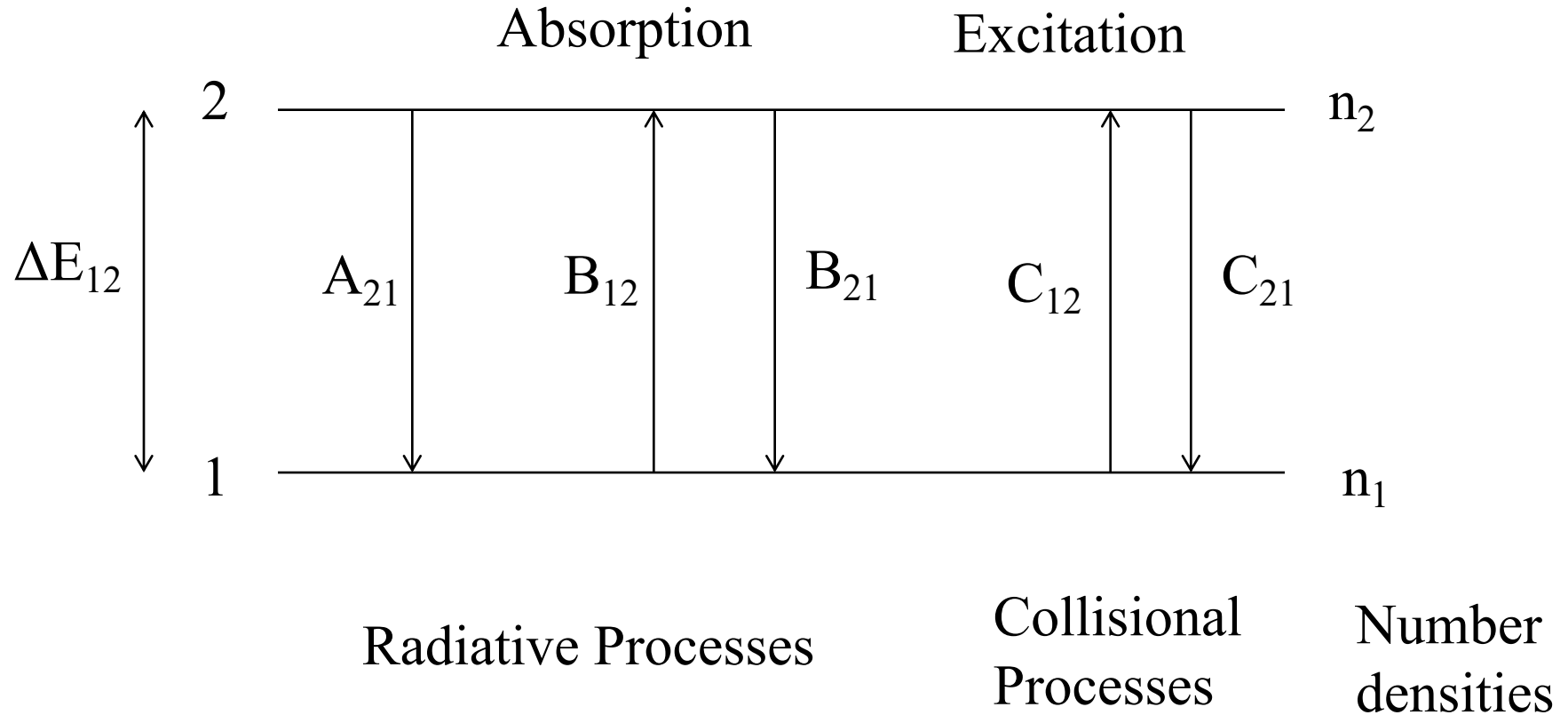


# Uses of Spectral Lines

- To determine the physical conditions such as density and temperature
- To determine abundances of species
- To probe the kinematics of the region
- To probe the magnetic field strength via the Zeeman effect

# Two-level system

Spontaneous Emission      Stimulated Emission      De-Excitation



# Einstein Coefficients

- $A_{21}$  is the probability that a spontaneous radiative decay from level 2 to level 1 per unit time ( $s^{-1}$ )
- $B_{12}$  is the coefficient for absorption of a photon during excitation from level 1 to level 2 ( $m^2 J^{-1} Hz st$ )
- $B_{21}$  is the coefficient for stimulated emission of a photon from level 2 to level 1

# Collision Coefficients

- $C_{12}$  is the collisional excitation coefficient describing the rate of collisional excitations from level 1 to level 2 ( $\text{m}^3\text{s}^{-1}$ )
- $C_{21}$  is the collisional de-excitation coefficient describing the rate of collisional de-excitations from level 2 to level 1

# Line Optical Depth

- Optical depth  $\tau_\nu$  is defined as

$$d\tau_\nu = \kappa_\nu ds$$

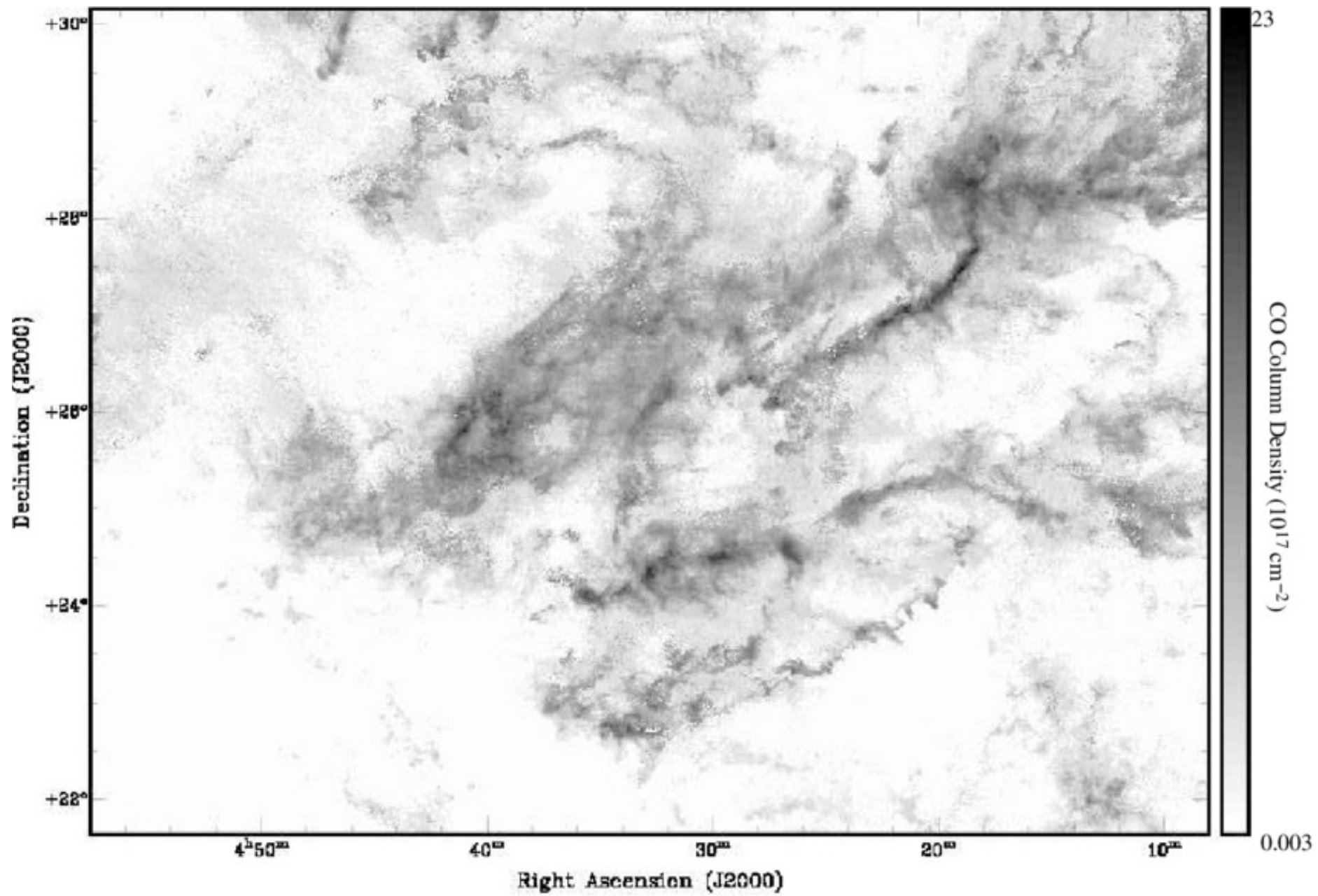
So for a spectral line

$$d\tau_\nu = (n_1 B_{12} - n_2 B_{21}) \frac{h\nu_{12}}{4\pi} \varphi(\nu) ds$$

# Column Density

- Note that the line optical depth is (mainly) proportional to  $n_l$
- This will be proportional to the total number density of the species,  $n$
- Hence,  $\tau_\nu \propto \int n ds \propto N$

where  $N$  is called the column density and has units of  $\text{m}^{-2}$



# Discussion

- What kind of assumptions would you need to make in order to estimate the number density,  $n$  ( $\text{m}^{-3}$ ), from measurements of the column density,  $N$  ( $\text{m}^{-2}$ )?

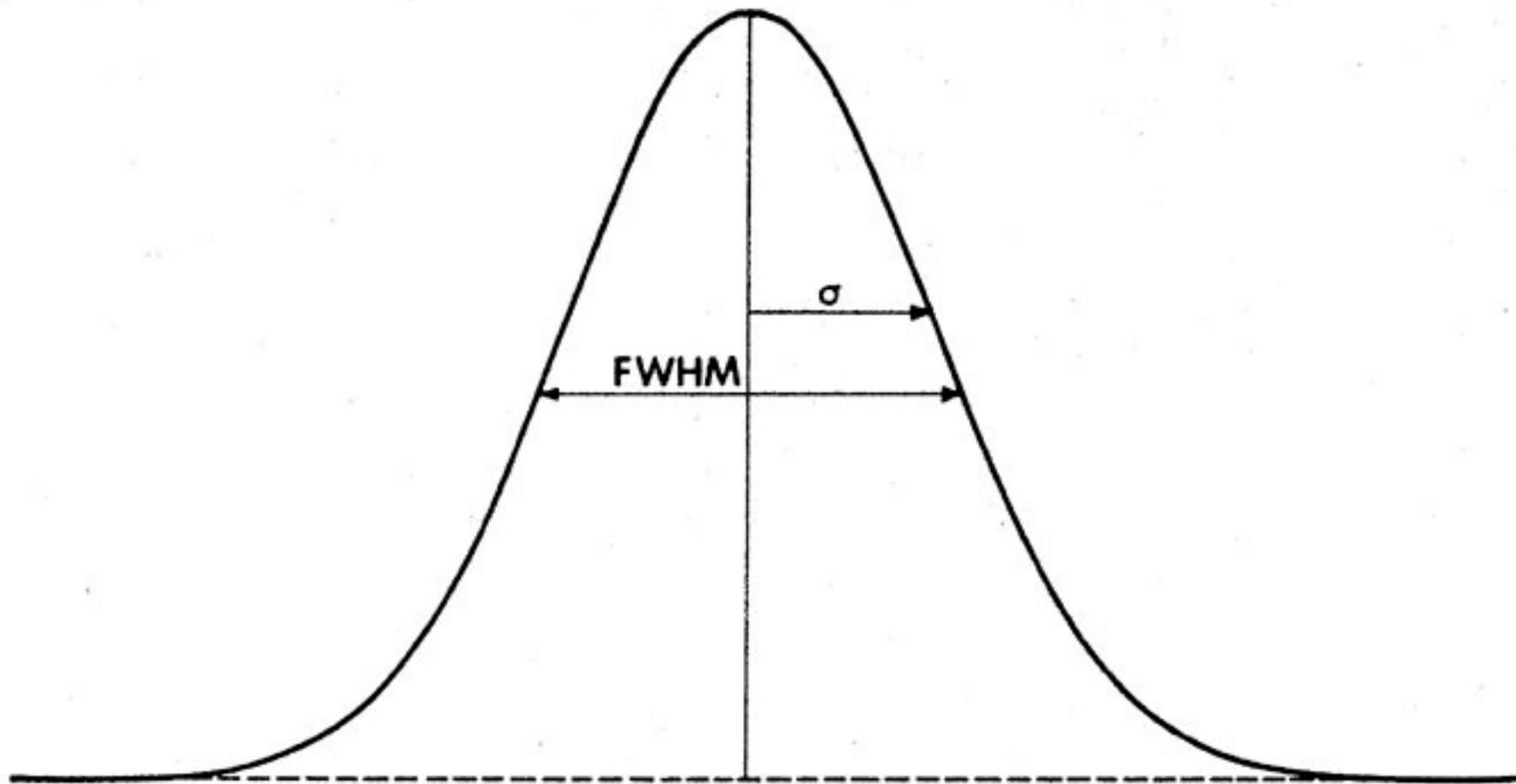
# Line Profile

For Doppler broadening by thermal motion of the molecules the frequency profile is Gaussian

$$\varphi(\nu) = \frac{1}{\sqrt{\pi} \Delta \nu_D} \exp - \frac{(\nu - \nu_0)^2}{\Delta \nu_D^2}$$

where the FWHM in velocity of the profile is

$$\Delta v_{FWHM} = 2\sqrt{\ln 2} \Delta v_D = 2\sqrt{\ln 2} \frac{\Delta \nu_D}{\nu_0} c = 2\sqrt{\ln 2} \left( \frac{2kT}{m} \right)^{\frac{1}{2}}$$



# Other Line Profiles

- Very often spectral lines are broader than their purely thermal width
- Usually due to Doppler shifts due to the kinematics of the gas, e.g.
  - Broad gaussian due to turbulence
  - Double-peaked due to rotation or outflow

# Negative Optical Depth

- In special circumstances the spectral line optical depth can go negative

$$d\tau_\nu = (n_1 B_{12} - n_2 B_{21}) \frac{h\nu_{12}}{4\pi} \varphi(\nu) ds$$

- This happens if  $n_2 > n_1$  – population inversion, and leads to maser amplification of the intensity as  $e^{\tau_\nu}$

# Line Emissivity

- We can write the line emissivity as

$$j_\nu = n_2 A_{21} \frac{h\nu}{4\pi} \varphi(\nu)$$

- Useful for interpreting optically thin line emission to derive column densities

# LTE

- In local thermodynamic equilibrium (LTE) each process is balanced by its inverse process, and the local temperature determines the level populations

$$j_\nu = \kappa_\nu B_\nu(T) \quad \text{or} \quad S_\nu = B_\nu(T)$$

- Good in optically thick situations and for continuum radiation like free-free

# Non-LTE

- Often LTE is not a good assumption
- Optically thin situations when the radiative rates are no longer determined by the local radiation field
- Need to solve full statistical equilibrium equations to determine level populations which can be characterized by an excitation temperature  $T_{\text{ex}} \neq T$

# Discussion

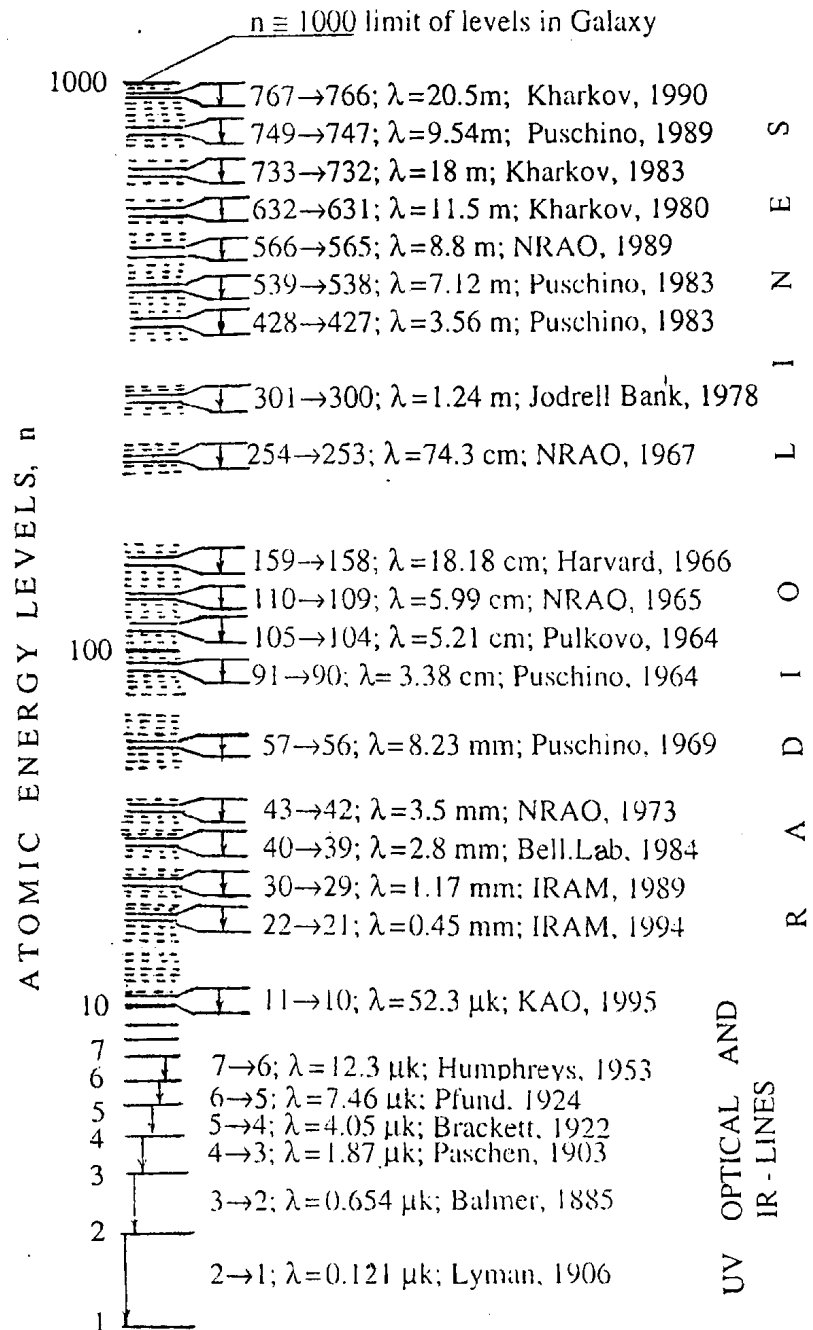
- Which of the following situations do you think LTE would be a good approximation?
  - The photosphere of Sun
  - The corona of the Sun
  - An H II region
  - The Big Bang

# Examples of Radio Lines

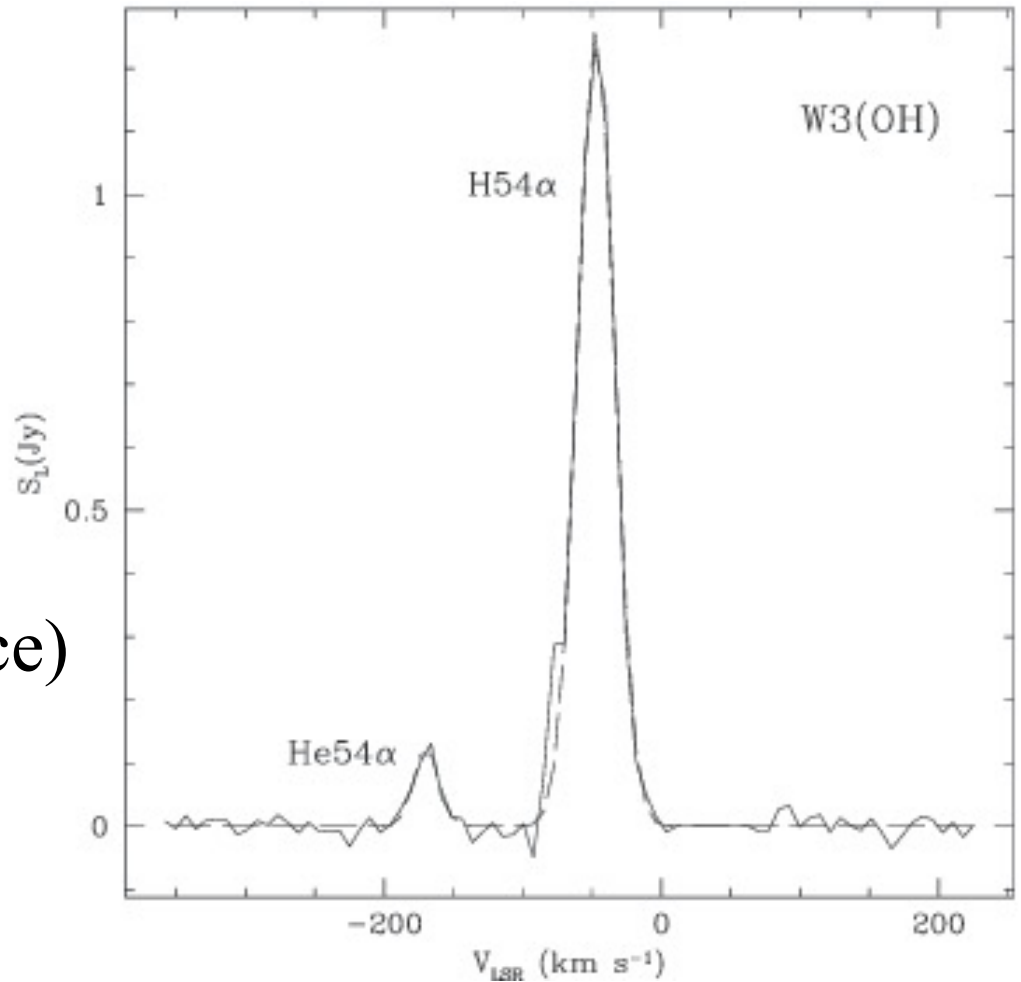
- There are spectral lines right across the radio spectrum
- These can arise from ions, atoms and molecules

# Radio Recombination Lines

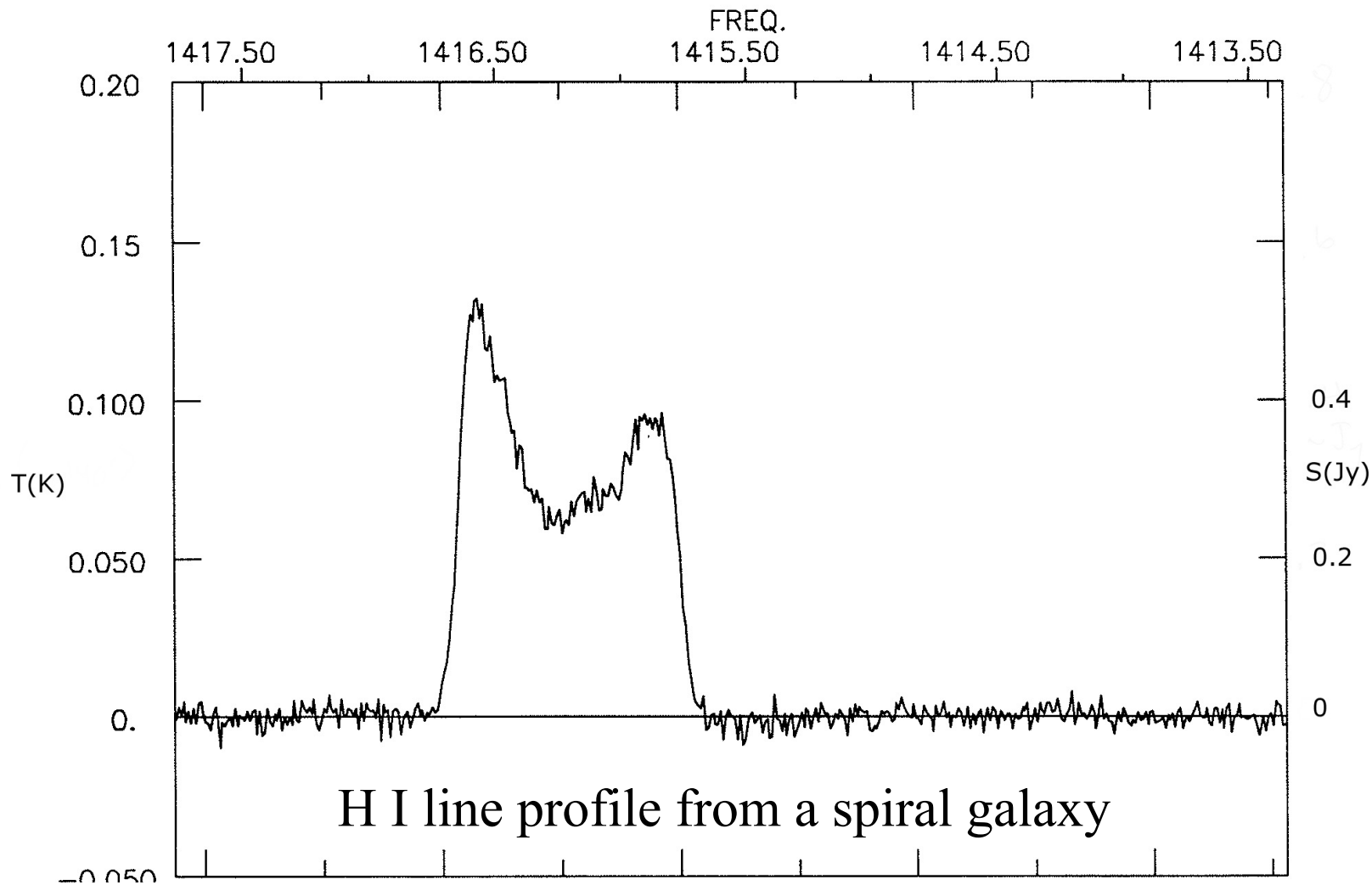
- Photo-ionized gas emits lines from electrons cascading down the high energy levels of hydrogen after recombination of a proton and electron

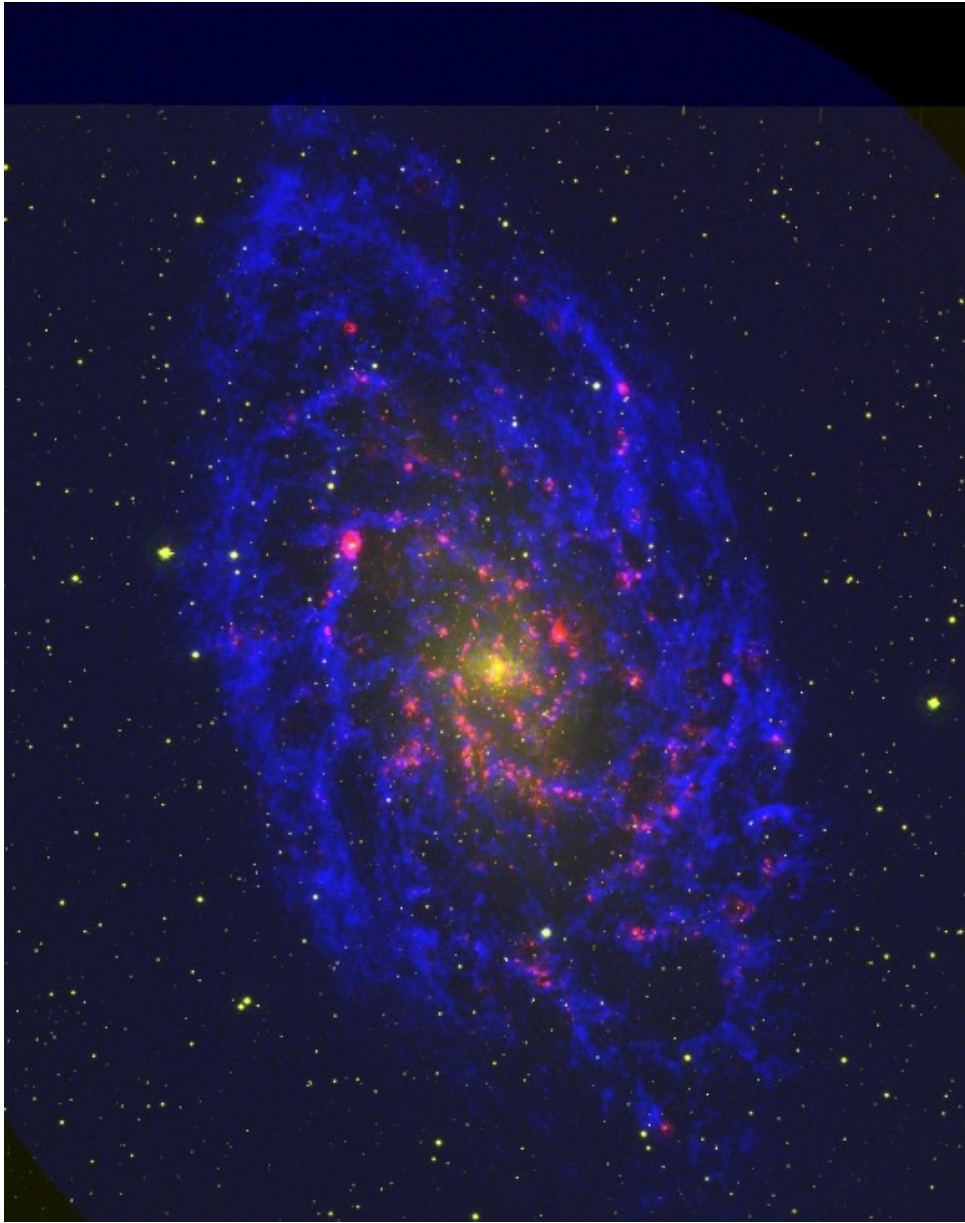


- Can use to diagnose:
  - Temperature
  - Density
  - He abundance
  - Velocity (Distance)
  - Expansion

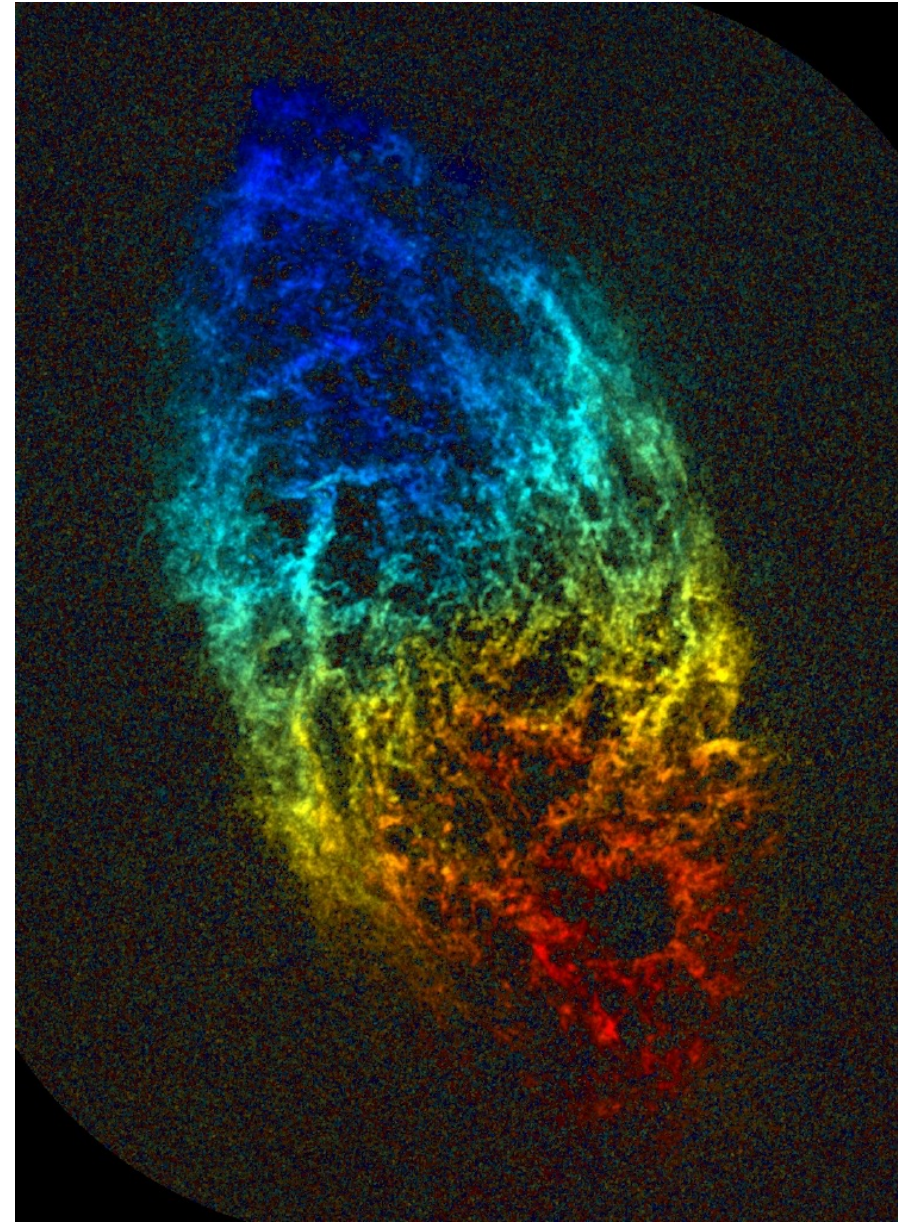


# H I 21 cm Line





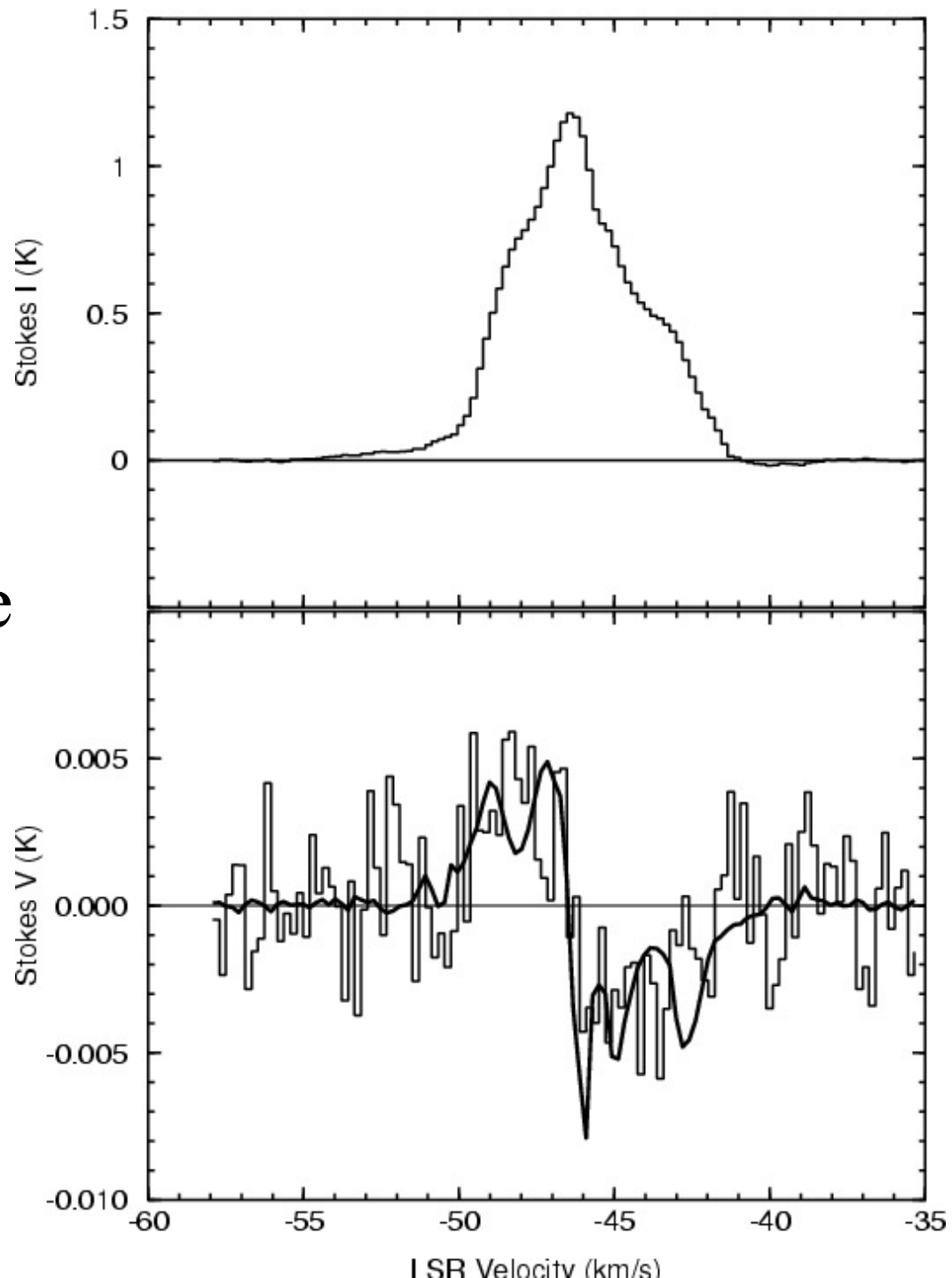
H I (blue) and optical image



Doppler shift (colours) of H I

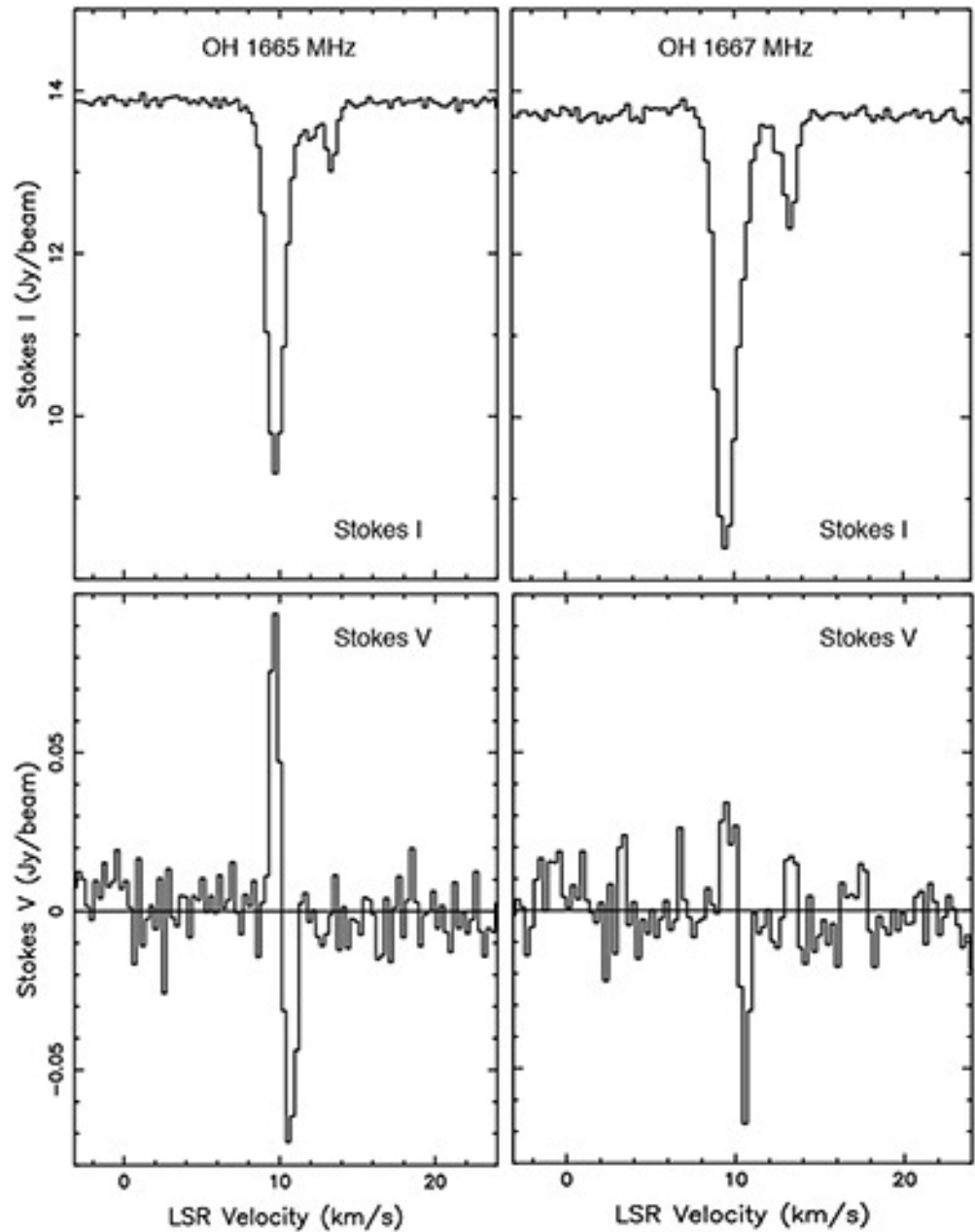
# Molecular Emission Lines

- E.g. CN molecule at 113 GHz
- Used to measure magnetic field via Zeeman effect



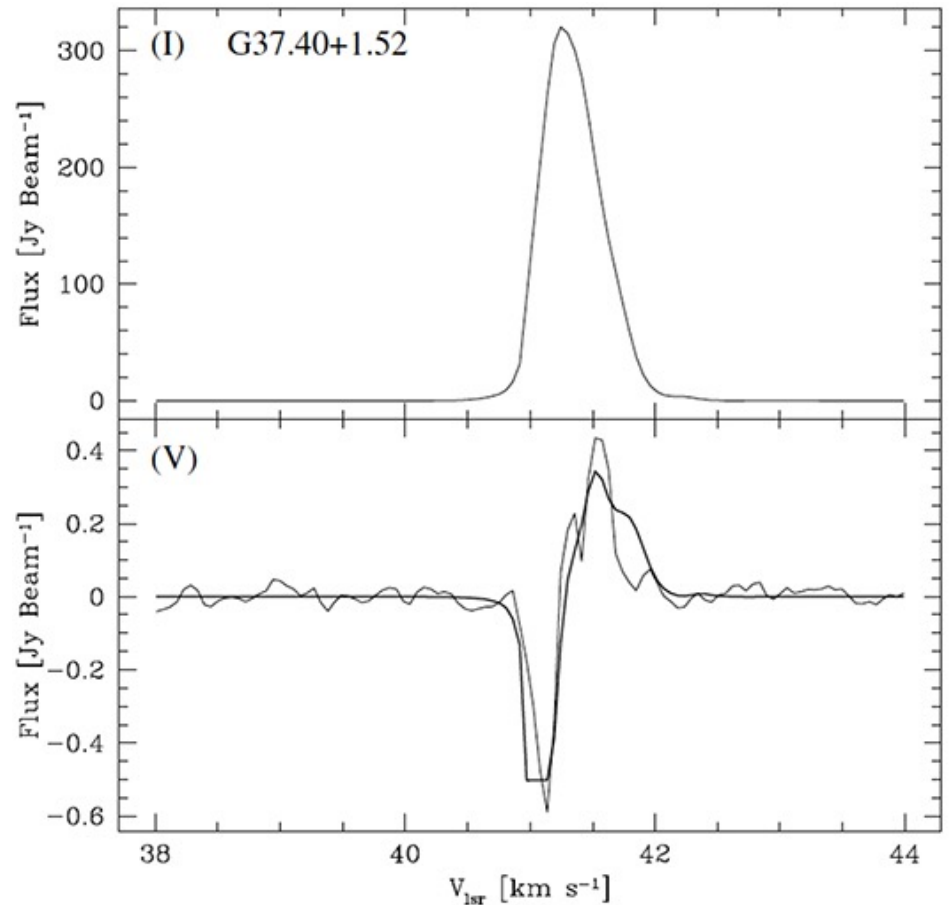
# Molecular Absorption Lines

- Lines appear in absorption against a strong background continuum source



# Maser Lines

- Several molecules exhibit strong non-thermal maser emission lines



# Summary

- Radio spectral lines can be used to determine:
  - Physical conditions
  - Abundances
  - Kinematics
  - Magnetic field